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A series of heterocyclic inhibitors of phenylalanyl-tRNA synthetases with antibacterial activity

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Abstract—A series of novel heterocyclic analogues have been synthesized and evaluated for their ability to inhibit phenylalanylt-RNA synthetases and act as antibacterial agents. Several analogues have good antibacterial activity against Staphylococcus aureus. © 2004 Elsevier Ltd. All rights reserved.

In the preceding paper,1 we described a series of spirocyclic furan and pyrrolidine inhibitors of Enterococci faecalis and Staphylococcus aureus phenylalanyl-tRNA synthetases (EfPheRS and SaPheRS). These compounds also exhibit Escherichia coli PheRS inhibitory activity. Based on lead compound 1a, generated by high throughput screening, a number of spirocyclic analogues were prepared. We discovered the isomeric compound 1b with potent activity against EfPheRS (IC₅₀ = 2 nM) and SaPheRS (IC₅₀ = 5 nM). Analogue 1b exhibited a relatively weak antibacterial activity (MIC = 50 μg/mL, S. aureus). Further study showed that both analogues 1a and 1b were unstable in Mueller Hinton Broth (MHB) bacterial media. The half-life time for 1a and 1b in MHB at 37°C is 117 min and 206 min respectively. To improve the stability and whole cell activity of this series, we sought to modify the structure of the molecule. Herein, we report the structure-activity relationship of a novel series of heterocyclic PheRS inhibitors with antibacterial activity against S. aureus.

It was found that derivatives generated by sodium borohydride reduction of spirocyclic analogues were stable in MHB. Scheme 1 shows the chemistry of these novel heterocyclic analogues. The preparation of spirocyclic compound 1b was described in the preceding paper.¹ Reduction² of analogue **1b** with sodium borohydride in methanol gave two isomers, which were purified by silica gel chromatography (15% ethyl acetate in dichloromethane). Elution of the column gave the slower moving isomer 3a and the faster moving isomer 3b. The ratio for isomer 3a to isomer 3b was 1:1.8. Both carbonyl groups on the spiro cyclic ring of compound 1b were reduced to the hydroxyl groups. Steric effects govern the stereochemistry of reduction. The ketone cis to the malimide ring is reduced selectively from less hindered face of the molecule to provide compounds 3a

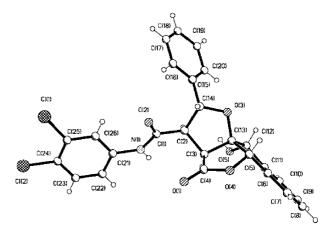


Figure 1. The crystal X-ray structure of analogue 3a.

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Scheme 1.

and **3b** where $H_{\rm d}$ is cis to the furan oxygen. The structure of these compounds was characterized by $^{\rm l}H$ NMR experiment and mass spectra. X-ray crystallography (Fig. 1) confirmed the stereochemistry. The protons $H_{\rm d}$ and $H_{\rm e}$ in analogue **3a** are cis to each other. Oxidation of the mixture of analogues **3a** and **b** with Dess–Martin reagent in acetonitrile generated carbonyl butyrolactone derivative **4**. Reduction of the cis spirocyclic analogue **1a** provided **5a–b**. Basic hydrolysis (NaOH/MeOH) of cis compound **1a** led to the maleimide ring open analogue **6**.

Heterocyclic analogues were evaluated for inhibition of the aminoacylation activity of EfPheRS and SaPheRS. Analogues were further tested for antibacterial activity against a panel of organisms. Table 1 shows the enzyme and whole cell activity. Both reduced hydroxyl butyrolactone derivatives $\bf 3a$ and $\bf b$ were less potent enzyme inhibitors than the parent compound $\bf 1b$. However, they each demonstrated a significant increase in antibacterial activity against *S. aureus*. Minimum inhibitory concentration (MIC) for analogues $\bf 3a$ and $\bf b$ was $\bf 3.1~\mu g/mL$ and $\bf 6.2~\mu g/mL$ respectively as compared with the corresponding compound $\bf 1b$ (MIC=50 $\mu g/mL$). Car-

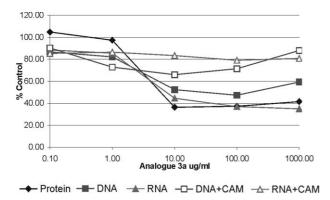


Figure 2. WT S. aureus macromolecular labeling with analogue 3a.

bonyl butyrolactone derivative **4** retained moderate enzyme activity, but had no activity against *S. aureus*. This result suggests the importance of the hydroxyl group in analogues **3a** and **b** to the antibacterial activity. For hydroxyl butyrolactone derivatives **5a** and **5b** from reduction of compound **1a**, only one isomer had the whole cell activity (MIC = $6.2 \mu g/mL$). The stereochemistry of **5a** and **b** was not assigned. Hydrolysis analogue **6** has moderate enzyme activity, but it was found to be inactive (MIC > $100 \mu g/mL$).

Inhibitors of *t*RNA synthetase activity produce a unique profile in assays monitoring macromolecular synthesis in *S. aureus*. As demonstrated in Figure 3, pseudomonic acid (PMA),^{4,5} a known inhibitor of isoleucine-*t*RNA synthetase, inhibits synthesis of protein, RNA, and DNA. RNA and DNA inhibition are due to induction of the stringent response.⁶ The addition of the protein synthesis inhibitor chloramphenicol (CAM) prevents induction of the stringent response, allowing RNA and DNA synthesis continues uninhibited until the dose of PMA is increased by 10–100-fold. As shown in Figure 2, analogue 3a has a similar dose–response profile to PMA. Inhibition of RNA synthesis is clearly

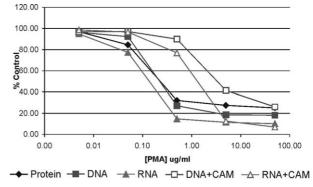


Figure 3. WT S. aureus macromolecular labeling with PMA.

Table 1. Inhibition of EfPheRS, SaPheR and minumn inhibitory concentration (MIC)

| Compd | Structure | Ic ₅₀ (μM) (EfPheRS) | $Ic_{50} (\mu M) (SaPheRS)$ | MIC (μg/mL) (S. aureus) |
|-----------------------|---|---------------------------------|-----------------------------|-------------------------|
| 3a | H OH H OH | 0.47 | 0.51 | 3.1 |
| 3b | HO H H O H H NH 3,4-Cl ₂ -Ph | 0.17 | 0.26 | 6.2 |
| 4 | O H Ph O H H H NH:3,4-Cl ₂ -Ph | 3.1 | 2.6 | > 100 |
| 5a ^a 5b | OH H Ph O = 1 Ph O NH 3,4-Cl ₂ -Ph | 15 4.9 | 14 >100 | > 100 63.2 |
| 6 | $\begin{array}{c} O & H \\ \hline \\ O & H \\ \hline \\ O & H \\ \end{array}$ $\begin{array}{c} O & H \\ \hline \\ O & H \\ \end{array}$ | 11 | 33 | > 100 |

^a The stereochemistry of the alcohol in **5a** and **5b** was not assigned.

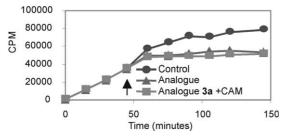


Figure 4. DNA Synthesis-analogue.

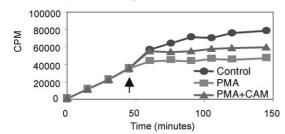


Figure 5. DNA Synthesis-PMA.

relieved by the addition of chloramphenicol, consistent with a phenylalanyl-tRNA synthetase as the primary target. Relief of DNA synthesis, however, was less striking, raising the possibility that the compound has an additional mode of action. This was confirmed by using a more sensitive kinetic assay to monitor DNA synthesis. As shown in Figure 5, addition of PMA results in an immediate cessation of DNA synthesis, whereas in the presence of chloramphenicol DNA synthesis continues for an additional 10 min, indicative of induction of the stringent response. In the presence of analogue 3a, however, DNA synthesis is inhibited immediately both with and without CAM (Fig. 4) indicating that DNA synthesis inhibition is due to activities other than inhibition of PheRS.

In conclusion, a novel structural class EfPheRS and SaPheRS heterocyclic inhibitors with in vitro antibacterial activity has been discovered. The best analogue 3a with activity against EfPheRS (IC₅₀=0.17 μ M)

and SaPheRS (IC₅₀=0.26 μ M) has potent antibacterial activity against *S. aureus* (MIC=3.1 μ g/mL).

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References and notes

- Yu, X. Y.; Finn, J.; Hill, J. M.; Wang, Z. G.; Keith, D.; Silverman, J.; Oliver, N. *Bioorg. Med. Chem. Lett.* 2004, 14, preceding paper in this issue. doi: 10.1016/ j.bmcl.2003.11.081.
- 2. Brown, H. C.; Muzzio, J. J. Am. Chem. Soc. 1966, 88, 2811.
- 3. Dess, D.; Martin, J. C. J. Am. Chem. Soc. 1991, 1138, 7277.
- Mellows, G. Int. Congr. Symp. Ser.-R. Soc. Med. 1984, 80, 23.
- 5. Hughes, J.; Mellows, G. Biochem. J. 1980, 191, 209.
- 6. Cashel, M. Encycl. Microbiol. (2nd Ed.) 2000, 4, 467.